

TETRAMETHYLAMMONIUM PERCHLORATE: MECHANISM OF ACTION IN COMPOSITIONS

V. P. Sinditskii, V. Yu. Egorshchey, A. O. Chepurnoy, A. N. Chernyi, A. A. Mikhaleva,
and E. V. Filonova

D. I. Mendeleev University of Chemical Technology of Russia, 9 Miusskaya Sq., Moscow 125047, Russian Federation

Abstract: The combustion behavior of compositions of tetramethylammonium perchlorate (TMAP) with inert additives, ammonium perchlorate (AP), and solid rocket propellants were studied. The reasons for the lack of combustion of TMAP in its pure form are explained. It turned out that the addition of Al_2O_3 to TMAP, due to higher thermal conductivity, leads to combustion increasing the heat feedback from the gas zone and compensating for the heat losses of the leading reaction in the condensed zone. With an increase in the proportion of Al_2O_3 or the use of nanotubes, the heat feedback from the gas phase increases so much that the combustion rate begins to be determined by the very rapid kinetics of decomposition of methyl perchlorate in the gas zone. On the example of the study of combustion of mixtures of TMAP with AP and propellants, it is shown that the efficiency of TMAP action depends on the mechanism of combustion of the composition: in the composition with the leading reaction in the condensed phase, TMAP plays the role of an inert additive and in compositions which burning rate is controlled by the gas-phase mechanism, TMAP plays the role of a fast heat-generating additive in the gas zone.

Keywords: energetic materials; perchlorates; tetramethylammonium perchlorate; combustion; combustion mechanism

DOI: 10.30826/CE24170210

EDN: GGBQUJ

Figure Captions

Figure 1 Dependence of combustion rate on pressure for mixtures of TMAP with Al_2O_3 (1 — TMAP/ Al_2O_3 50/50; 2 — 70/30; and 3 — TMAP/ Al_2O_3 80/20) and nanotubes (CNTs) (4 — TMAP/CNTs 90/10); dashed line — estimated burning rate of TMAP

Figure 2 Comparison of surface temperatures of AP (dashed line) and the TMAP/20% Al_2O_3 mixture (solid line): 1 — at pressure of 5 atm; 2 — from data on combustion rates; and 3 — decomposition temperature

Figure 3 Dependence of burning rate on pressure of binary AP/TMAP compositions containing 5% and 10% TMAP

Figure 4 Dependence of burning rate on pressure of binary AP/TMAP compositions containing 20% and 30% TMAP

Figure 5 Dependence of combustion rate on pressure of propellants based on polyetherurethane polymer (1) with TMAP additives: 2 — 60%; 3 — 40%; 4 — 20%; 5 — 10%; and 6 — 5%

Figure 6 Dependence of combustion rate on pressure of propellants containing (black curves) and not containing (grey curve) ferrocene catalyst with 20% TMAP additives (solid curves)

References

1. Fogelzang, A. E., and B. S. Svetlov. 1970. O svyazi mezhdu strukturoy vzryvchatykh veshchestv i skorost'yu ikh goreniya [On the relationship between the structure of explosives and their burning rate]. *Dokl. Akad. Nauk SSSR* 192(6):1322–1325.
2. Fogel'zang, A. E., B. S. Svetlov, V. S. Opryshko, and V. Ya. Adzhemyan. 1972. Investigation of the combustion of organic perchlorates. *Combust. Explo. Shock Waves* 8(2):206–217. doi: 10.1007/BF00740449.
3. Sinditskii, V. P., V. Yu. Egorshchey, V. V. Serushkin, A. O. Chepurnoy, and A. A. Mikhaleva. 2023. Mekhanizm goreniya perkhloratov metil'nykh proizvodnykh ammoniya [Combustion mechanism of methyl ammonium perchlorates]. *Goren. Vzryv (Mosk.) — Combustion and Explosion* 16(3):93–99.
4. Inami, S. E., W. A. Rosser, and B. Wise. 1963. Dissociation pressure of ammonium perchlorate. *J. Phys. Chem.* 67:1077–1079.
5. Rosen, J. M., and C. Dickenson. 1969. Vapor pressures and heats of sublimation of some high melting organic explosives. *J. Chem. Eng. Data* 14:120–124.
6. Cundall, R. B., T. F. Palmer, and C. E. C. Wood. 1978. Vapor pressures measurements of some organic explosives. *J. Chem. Soc. Farad. T.* 174:1339–1345.
7. Jain, S. R., K. C. Adiga, V. R. Pai Verneker. 1979. Combustion of ammonium perchlorate-based composite propellants in presence of methylammonium perchlorates. *Combust. Flame* 35:225–231.

8. Novozhilov, B. V. 1973. *Nestatsionarnoe gorenje tverdogo raketnogo topliva* [Nonstationary combustion of solid rocket propellants]. Moscow: Nauka. 176 p.
9. Denisyuk, A. P., L. A. Demidova, and V. I. Galkin. 1995. The primary zone in the combustion of solid propellants containing catalysts. *Combust. Explo. Shock Waves* 31(2):161–167. DOI: 10.1007/BF00755743. EDN: WUVLIF.
10. Balandin, A. A. 2011. Thermal properties of graphene and nanostructured carbon materials. *Nat. Mater.* 10(8):569–581.
11. Kondrashkov, Y. A., N. N. Bakhman, and A. F. Belyaev. 1967. Burning rate of composite systems as a function of the particle size of the components at different fuel-oxidizer ratios. *Combust. Explo. Shock Waves* 3(3):210–212. doi: 10.1007/BF00791862.
12. Sinditskii, V. P., A. N. Chernyi, and D. A. Marchenkov. 2012. Issledovanie gorenija topliv na osnove perkhlorata ammoniya s nizkim koeffitsientom izbytkha okislitelya [Combustion study of ammonium perchlorate-based propellants with low oxidizer-to-fuel ratio]. *Khimicheskaya fizika i mezoskopiya* [Chemical Physics and Mesoscopy] 14(4):519–524.
13. Sinditskii, V. P., V. Yu. Egorshhev, A. N. Chernyi, et al. 2011. Zakonomernosti i mekhanizm gorenija perkhlorata ammoniya i ego smesey s aktivnym svyazuyushchim [Regularities and mechanism of combustion of ammonium perchlorate and its mixtures with an active binder]. *Goren. Vzryv (Mosk.) — Combustion and Explosion* 4:236–242.
14. Meyer, J., and W. Spormann. 1936. Zur Kenntnis der Ester der Überchlorsäure. *Z. Anorg. Allg. Chem.* 228(4):341–351.

Received December 20, 2023

Contributors

Sinditskii Valeriy P. (b. 1954) — Doctor of Science in chemistry, professor, dean, Chemical Engineering Department, D. I. Mendeleev University of Chemical Technology of Russia, 9 Miusskaya Sq., Moscow 125047, Russian Federation; vps@muctr.ru

Egorshhev Vyacheslav Yu. (b. 1959) — senior lecturer, Department of Chemistry and Technology of Organic Nitrogen Compounds, D. I. Mendeleev University of Chemical Technology of Russia, 9 Miusskaya Sq., Moscow 125047, Russian Federation; egorshhev@yahoo.com

Chepurnoy Aleksey O. (b. 1992) — engineer, Department of Chemistry and Technology of Organic Nitrogen Compounds, D. I. Mendeleev University of Chemical Technology of Russia, 9 Miusskaya Sq., Moscow 125047, Russian Federation; chepurnoi.a.o@muctr.ru

Chernyi Anton N. (b. 1985) — teacher, Department of Chemistry and Technology of Organic Nitrogen Compounds, D. I. Mendeleev University of Chemical Technology of Russia, 9 Miusskaya Sq., Moscow 125047, Russian Federation; chernyi.a.o@muctr.ru

Mikhaleva Alena A. (b. 1996) — student, Department of Chemistry and Technology of Organic Nitrogen Compounds, D. I. Mendeleev University of Chemical Technology of Russia, 9 Miusskaya Sq., Moscow 125047, Russian Federation; mixal1996@yandex.ru

Filonova Ekaterina V. (b. 2001) — student, Department of Chemistry and Technology of Organic Nitrogen Compounds, D. I. Mendeleev University of Chemical Technology of Russia, 9 Miusskaya Sq., Moscow 125047, Russian Federation; 181296@muctr.ru