STUDY OF PENTAERYTHRITOL TETRANITRATE SUBLIMATION BY ATOMIC FORCE MICROSCOPY

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Abstract: Sublimation of energetic materials is a critical process for their application and detection. Studies of sublimation using atomic force microscopy (AFM) allow one to monitor changes in the volume of individual microparticles and thus analyze compounds which are not suitable for classical research being too dangerous in handling or decomposing before sublimation under ambient conditions. However, both sample morphology and experimental conditions could significantly alter the results of the AFM study. The authors have studied in detail the effect of these factors alone with the substrate material influence on the correct assessment of the enthalpy of sublimation using AFM. The authors have observed the mechanical influence of the AFM probing process itself and the thermally-induced morphological changes of pentaerythritol tetranitrate (PETN) particles being heated at probing. Simultaneously with the PETN particles sublimation, at 45-60 °C, the increase in the particles volume was found indicating the recrystallization process which distorts the measurement results. However, on the mica substrate, the recrystallization process was not observed and the enthalpy of PETN sublimation could be correctly estimated from AFM. Additionally, it was found that for the correct measurement of the sublimation enthalpy of PETN, the particle height should exceed 280 ± 30 nm. In general, very small sample masses (< 10 μ g) and relatively low temperatures used in AFM analysis allow safely obtain the correct sublimation enthalpy data for highly sensitive or low-volatile thermally stable materials and for the newly synthesized in small quantities energetic compounds with unknown properties.

Keywords: atomic-force microscopy; scanning probe microscopy; sublimation; PETN; energetic materials

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Figure Captions

Figure 1 Thermally-induced morphological changes of PETN on Al₂O₃ substrate

Figure 2 Changes of volume (a) and volume change rate constant k (b) with temperature for PENT on mica (filled signs) and gold substrates (empty signs)

Figure 3 Mechanically-induced morphological changes of PETN on Si substrate: (*a*) crystallization of amorphous PETN in the scan area; and (*b*) destruction of the PETN particles and displacement of the fragments to the edges of the scan area

Figure 4 Changes of the PETN sublimation enthalpy estimated by AFM with material of the substrate (a) and particle height (b)

Table Captions

Table 1 Morphology and average size of PETN particles d_m on different substrates

Table 2 Experimentally measured average roughness R_a and contact angle θ and literature data on heat capacity C_p at 298,15 K and thermal conductivity λ of the substrates

 Table 3 Literature data on PETN sublimation enthalpy

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References

- 1. Ewing, R. G., M. J. Waltman, D. A. Atkinson, J. W. Grate, and P. J. Hotchkiss. 2013. The vapor pressures of explosives. *TRAC Trend. Anal. Chem.* 4:35–48.
- Inozemtsev, Y. O., A. B. Vorobjov, A. V. Inozemtsev, and Yu. N. Matyushin. 2014. Kalorimetriya energoemkikh soedineniy [Calorimetry of energetic materials]. *Goren. Vzryv (Mosk.) – Combustion and Explosion* 7:260–270.
- 3. Price, D. M. 2015. A fit of the vapours. *Thermochim. Acta* 622:44–50.
- 4. Verevkin, S. P., D.H. Zaitsau, V. N. Emel'yanenko, and A. A. Heintz. 2011. A new method for the determination of vaporization enthalpies of ionic liquids at low temperatures. *J. Phys. Chem. B* 115(44):12889–12895.
- Burnham, A. K., S. R. Qiu, R. Pitchimani, and B. L. Weeks. 2009. Comparison of kinetic and thermodynamic parameters of single crystal pentaerythritol tetranitrate using atomic force microscopy and thermogravimetric analysis: Implications on coarsening mechanisms. *J. Appl. Phys.* 105(10):104312.
- Pitchimani, R., A. K. Burnham, and B. L. Weeks. 2007. Quantitative thermodynamic analysis of sublimation rates using an atomic force microscope. *J. Phys. Chem. B* 111(31):9182–9185.
- 7. Mridha, S., and B. L. Weeks. 2009. Effect of Zn doping on the sublimation rate of pentaerythritol tetranitrate using atomic force microscopy. *Scanning* 31(5):181–187.
- 8. Tung, Y. S., R. Mu, A. Ueda, D. O. Henderson, W. A. Curby, and A. Mercado. 2002. The study of sublimation rates and nucleation and growth of TNT and PETN on slica and graphite surfaces by optical and atomic force microscopy and ellipsometry. *Atomic force microscopy/scanning tunneling microscopy 3.* Eds. S. H. Cohen and M. L. Lightbody. New York, NY: Springer. 135–152.
- 9. Lee, Y.J., and B. L. Weeks. 2019. Investigation of sizedependent sublimation kinetics of 2,4,6-trinitrotoluene

(TNT) micro-islands using in-situ atomic force microscopy. *Molecules* 24(10):1895.

- Kuchurov, I. V., I. V. Fomenkov, S. G. Zlotin, and V. A. Tartakovsky. 2012. Synthesis of nitric acid esters from alcohols in a dinitrogen pentoxide/carbon dioxide liquid system. *Mendeleev Commun.* 22(2):67–69.
- 11. Nečas, D., and P. Klapetek. 2012. Gwyddion: An opensource software for SPM data analysis. *Open Phys.* 10(1).
- Muravyev, N. V., K. A. Monogarov, I. N. Melnikov, A. N. Pivkina, and V. G. Kiselev. 2021. Learning to fly: Thermochemistry of energetic materials by modified thermogravimetric analysis and highly accurate quantum chemical calculations. *Phys. Chem. Chem. Phys.* 23(29):15522–15542.
- Fondren, Z. T., N. S. Fondren, G. B. McKenna, and B. L. Weeks. Crystallization kinetics of pentaerythritol tetranitrate (PETN) thin films on various materials. 2020. *Appl. Surf. Sci.* 522:146350.
- Zhang, G., and B. L. Weeks. 2008. Inducing dendrite formation using an atomic force microscope tip. *Scanning* 30(3):228–231.
- 15. Lide, D. R., ed. 2014–2015. *CRC handbook of chemistry and physics*. 85th ed. CRC Press. 2661 p.
- Cady, H. H., and A. C. Larson. 1975. Pentaerythritol tetranitrate II: its crystal structure and transformation to PETN I: An algorithm for refinement of crystal structures with poor data. *Acta Crystall. B – Stru.* 31(7):1864–1869.
- Östmark, H., S. Wallin, and H. G. Ang. 2012. Vapor pressure of explosives: A critical review. *Propell. Explos. Pyrot.* 37(1):12–23.
- Hikal, W. M., J. T. Paden, and B. L. Weeks. 2011. Simple method for determining the vapor pressure of materials using UV-absorbance spectroscopy. *J. Phys. Chem. B* 115(45):13287–13291.

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