

THERMAL EXPLOSION OF NEPE-TYPE COMPOSITE ENERGETIC MATERIAL

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Abstract: The times to thermal explosion τ of a mixed energetic material of the NEPE (Nitrate Ester Plasticized PolyEster) type were experimentally determined on samples in the form of cylinders with a diameter and a height of 4 mm, pressed into hermetic steel shells with a filling density close to theoretical. The experiments were carried out under isoperibolic conditions in the temperature range from 110 to 140 °C. The obtained experimental values of τ are compared with those calculated by the system of equations of thermal conductivity and kinetics using the kinetic characteristics obtained by the DSC (differential scanning calorimetry) method. It is shown that the results of experiments and numerical calculations are in the best agreement when using the formal kinetic equation of a branched chain reaction.

Keywords: thermal explosion; thermal decomposition; energetic material; nitrate esters; activation energy

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Figure Captions

Figure 1 Normalized DSC peaks of the plasticizer in the structure of material K-1 obtained in sealed crucibles (solid lines) and dependences $d\alpha/dT$ at heating rates $b = 0.1, 0.35$, and 1.0 K/min (from left to right) and values $j = 0.99999$ (1–3), and 0.999 (1'–3')

Figure 2 Dependences of the rate of thermal decomposition of plasticizer of material K-1 on time calculated by Eq. (1) for $T = 120$ °C at $j = 0.999$ (1), 0.9999 (2), and 0.99999 (3). Solid lines — $n = 0, m = 1$; and dotted lines — $n = 1, m = 1$

Figure 3 Dependence of temperature in the center of a polymethyl methacrylate cylinder on time when the cylinder is placed in a thermostat preheated to 100 °C: signs — experiment; and curves — calculations at $H = 5$ (1), 10 (2), and 20 W/(m²·K) (3)

Figure 4 Calculated dependences of the temperature in the center of the sample of material K-1 in the form of a ball on time at $T_s = 120$ °C: 1–3 — $F(T, t)$ according to Eq. (4) (first-order autocatalysis) at $\alpha_0 = 0.001$ (1), 0.0001 (2), and 0.00001 (3); and 4 — $F(T, t)$ according to Eq. (3) at $j = 0.999$

Figure 5 Dependence of the delay time of thermal explosion of material samples K-1 on the ambient temperature: 1 and 2 — calculation according to Eq. (2) with initial and boundary conditions (5) and (6) using the heat release function (3) (1 — $j = 0.9999$ and 2 — $j = 0.999$); 3 — experimental data obtained in the explosive cells described above; and 4 — DSC data with samples weighing 1–2 mg

Figure 6 Experimental dependence of the delay time of thermal explosion of K-1 samples on temperature (signs) and its approximation by the function $y = K \exp[E/(Rx)]$ (curve)

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