

IGNITION OF GAS MIXTURE BY COMBUSTION PRODUCTS OF THERMITE COMPOSITION Al/CuO

B. D. Yankovskii¹, S. Yu. Anan'ev¹, A. Yu. Dolgoborodov^{1,2,3}, L. I. Grishin^{1,3}, and G. S. Vakorina¹

¹Joint Institute for High Temperatures of the Russian Academy of Sciences, 13-2 Izhorskaya Str., Moscow 125412, Russian Federation

²N. N. Semenov Federal Research Center for Chemical Physics of the Russian Academy of Sciences, 4 Kosygin Str., Moscow 119991, Russian Federation

³National Research Nuclear University MEPhI, 31 Kashirskoe Sh., Moscow 115409, Russian Federation

Abstract: The paper presents new experimental results on the dynamics of the cloud of explosive combustion products of the mechanoactivated composition Al/CuO. The parameters of the cloud of combustion products depending on the mass of the mixture were determined using the methods of high-speed photoregistration, pyrometric measurements, and photovoltaic and electrocontact sensors. Various methods of ignition and formation of the product flow are considered. Optimal conditions for the formation of a torch for ignition of combustible gas–air mixtures have been determined.

Keywords: aluminum; copper oxide; mechanical activation; explosive combustion; ignition; combustible gas

DOI: 10.30826/CE22150109

Figure Captions

Figure 1 Schemes of experimental assemblies for the formation of a thermite mixture combustion torch: 1 — a weighed portion of the thermite mixture; 2 — point of initiation; 3 — cloud of products; 4 — electrical contact sensor; 5 — channel; 6 — target; M_{cm} — mass of the sample; E_{init} — spark energy; $d_{channel}$ — channel diameter; $L_{channel}$ — channel length; l_{init} — distance from the initiation point to the open end of the channel; and L_{mix} — length of the mixture charge in the channel

Figure 2 A typical photograph (a) and graphical representation of the expansion dynamics of the luminous area during burning of a mixture sample in free space: 1 — 0.06 g; 2 — 0.25; 3 — 0.75, and 4 — 1.5 g (b)

Figure 3 Expansion dynamics of the luminous area in time: 1 — 0.06 g; 2 — 0.25; 3 — 0.75, and 4 — 1.5 g

Figure 4 Rate of expansion of the luminous area depending on the weight of the sample: 1 — $dV/dt = -0,6M^2 + 2,1M$; and 2 — $dV/dt = 1,5M$

Figure 5 A typical photograph of a quasi-cylindrical torch of mixture combustion (0.06 g). The mixture sample was placed in a shell with a depth of 2 mm with one free surface (a); and dynamics of linear expansion (b) and volume increase (c) of the luminous area at different energies of electrospark initiation: 1 — 120 mJ; and 2 — 20 mJ

Figure 6 Dynamics of expansion of the luminous area (a) and the increase in the torch volume (b) depending on the location of initiation point along the channel depth

Figure 7 The breakdown-burning traces of the flow of reacting mixture particles on a 0.3-millimeter thick polymer target

Figure 8 Photographs of ignition process of propane–butane gas mixture: (a) the initial stage of the initiation process in the chamber; and (b) radiation recorded outside the chamber with the release and afterburning of propane–butane–air mixture outside the chamber

Table Caption

The results for the dynamics of the flame region for various schemes of mixture location

Acknowledgments

Experimental results on the parameters of initiation and combustion were carried out at the Joint Institute for High Temperatures of the Russian Academy of Sciences within the framework of a state assignment for the development of fundamental principles for obtaining new materials. Mechanically activated mixtures were prepared at N. N. Semenov Federal Research Center for Chemical Physics of the Russian Academy of Sciences within the

framework of the state assignment for fundamental research of a new generation energetic materials. During the experimental studies, the equipment of the Explosive Center for Collective Use of the Russian Academy of Sciences and the unique scientific installation “Sphere” have been used.

References

1. Dolgoborodov, A. Yu. 2015. Mechanically activated oxidizer–fuel energetic composites. *Combust. Explo. Shock Waves* 51(1):86–99.
2. Dreizin E. L., and M. Schoenitz. 2017. Mechanochemically prepared reactive and energetic materials: A review. *J. Mater. Sci.* 52(20):11789–11809.
3. Streletskii, A. N., M. V. Sivak, and A. Yu. Dolgoborodov. 2017. Nature of high reactivity of metal/solid oxidizer nanocomposites prepared by mechanoactivation: A review. *J. Mater. Sci.* 52(20):11810–11825.
4. Dolgoborodov, A. Yu., V. G. Kirilenko, A. N. Streletskii, I. V. Kolban'ev, A. A. Shevchenko, B. D. Yankovskii, S. Y. Anan'ev, and G. E. Val'yano. 2018. Mechanoactivated thermite composition Al/CuO. *Goren. Vzryv (Mosk.) — Combustion and Explosion* 11(3):117–124.
5. Polak, L. S., A. A. Ovsyannikov, D. I. Slovetsky, and F. B. Wurzel. 1975. *Teoreticheskaya i prikladnaya plazmokhimiya* [Theoretical and applied plasma chemistry]. Moscow: Nauka. 304 p.
6. Kondratiev, V. N., and E. E. Nikitin. 1981. *Khimicheskie protsessy v gazakh* [Chemical processes in gases]. Moscow: Higher School. 264 p.
7. Lawton, J., and F. J. Weinberg. 1969. *Electrical aspects of combustion*. Oxford, U.K.: Clarendon Press. 419 p.
8. Anan'ev, S. Y., A. Yu. Dolgoborodov, and B. D. Yankovskii. 2017. Dinamika razleta produktov goreniya mekhanoaktivirovannoy smesi alyuminiya s oksidom medi [Expansion dynamics of combustion products of a mechanically activated mixture of aluminum with copper oxide]. *Goren. Vzryv (Mosk.) — Combustion and Explosion* 10(4):81–85.
9. Bratton, K. R., C. Woodruffa, L. L. Campbell, R. J. Heaps, and M. L. Pantoya. 2020. A closer look at determining burning rates with imaging diagnostics. *Opt. Laser. Eng.* 124:105841.

Received December 2, 2021

Contributors

Yankovskii Boris D. (b. 1947) — Candidate of Science in physics and mathematics, senior research scientist, Joint Institute for High Temperatures of the Russian Academy of Sciences, 13-2 Izhorskaya Str., Moscow 125412, Russian Federation; yiy2004@mail.ru

Anan'ev Sergey Yu. (b. 1990) — Candidate of Science in physics and mathematics, research scientist, Joint Institute for High Temperatures of the Russian Academy of Sciences, 13-2 Izhorskaya Str., Moscow 125412, Russian Federation; serg.anan'ev@gmail.com

Dolgoborodov Alexander Yu. (b. 1956) — Doctor of Science in physics and mathematics, chief research scientist, N. N. Semenov Federal Research Center for Chemical Physics of the Russian Academy of Sciences, 4 Kosygin Str., Moscow 119991, Russian Federation; head of laboratory, Joint Institute for High Temperatures of the Russian Academy of Sciences, 13-2 Izhorskaya Str., Moscow 125412, Russian Federation; teacher, National Research Nuclear University MEPhI, 31 Kashirskoe Sh., Moscow 115409, Russian Federation; aldol@ihed.ras.ru

Grishin Leonid I. (b. 1993) — junior research scientist, Joint Institute for High Temperatures of the Russian Academy of Sciences, 13-2 Izhorskaya Str., Moscow 125412, Russian Federation; Ph.D. student, National Research Nuclear University MEPhI, 31 Kashirskoe Sh., Moscow 115409, Russian Federation; lenya-grishin@mail.ru

Vakorina Galina S. (b. 1978) — Candidate of Science in physics and mathematics, leading engineer, Joint Institute for High Temperatures of the Russian Academy of Sciences, 13-2 Izhorskaya Str., Moscow 125412, Russian Federation; vakorinags@ihed.ras.ru