

MEASUREMENT OF THE HEATS OF COMBUSTION OF NATURAL COMBUSTIBLE GASES IN THE COMBUSTION BOMB CALORIMETER

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Abstract: A bomb calorimeter design for measuring the calorific values of combustible gases is proposed and tested. The energy equivalent of calorimeter does not depend on the standard substance burned — gas (high-purity methane) or benzoic acid. The sources of the error of calorimetric measurements are determined and design solutions are proposed that allow obtaining accurate results. Data on the measurements are provided, a comparison of the obtained values confirms the correctness of the technical solutions.

Keywords: bomb calorimeter; volumetric heat of combustion; calorific value; energy equivalent; measurement error

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