THE EFFECT OF HYDROGEN PEROXIDE ON NITRIC OXIDE FORMATION AT COMBUSTION OF THE VAPOR/AIR/METHANE MIXTURES

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Abstract: The effect of the hydrogen peroxide (H₂O₂) on the formation of nitric oxide (NO) at combustion of H₂O/air/CH₄ mixture is studied. It is demonstrated that the addition of hydrogen peroxide to the stoichiometric mixture of H₂O/air/CH₄ increases the yield of NO. The observed increase is mainly due to the overstoichiometric amount of oxygen introduced to the mixture by addition of the hydrogen peroxide as well as to the small (7 K), but noticeable increase in the value of maximum temperature reached during the combustion of methane (T_{max}). The observed increase in the value of T_{max} is due to the higher heat release of the reaction $3H_2O_2 + CH_4 = 5H_2O + CO (\Delta_r H_{298.15}^0(1) = -836.3 \text{ kJ/mol CH}_4)$ than that for the reaction $1.5O_2 + CH_4 = 2H_2O + CO (\Delta_r H_{298.15}^0(2) = -519 \text{ kJ/mol CH}_4)$.

Keywords: CH₄; H₂O₂; combustion; NO

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