INFLUENCE OF MECHANICAL ACTIVATION ON IGNITION OF ARESTED REACTIVE MILLING AI/MoO₃ COMPOSITES

M. V. Sivak¹, A. N. Streletskii^{1,2}, A. A. Shevchenko¹, and I. V. Kolbanev¹

¹N. N. Semenov Institute of Chemical Physics, Russian Academy of Sciences, 4 Kosygin Str., Moscow 119991, Russian Federation

²Moscow Institute of Physics and Technology (State University), 9 Institutskiy Per., Dolgoprudny, Moscow Region 141701, Russian Federation

Abstract: The factors affecting the ability of Arested Reactive Milling (ARM) nanocomposites Al/MoO₃ to be ignited by the hot substrate were analyzed. For this purpose, the composites using Al with different particle sizes (micro or nano) and morphology (spherical or laminar) were prepared. Also, the concentration of defects in MoO₃ [N] and the value of contact surface S_c between ingredients were controlled. There are two critical parameters: T_0 — the temperature of the substrate, at which the ignition delay time (τ) < 1 s; and T_{∞} — the temperature of the substrate, at which the ignition delay time (τ) depends linearly on substrate temperature. It is shown that T_{∞} and T_0 decrease with the contact surface area and correlate with the concentration of paramagnetic centers in MoO₃. The minimum values of $T_0 = 355$ °C and $T_{\infty} = 265$ °C were found for ARM composites, prepared from lamellar aluminum particles. In this case, low ignition temperatures can be attained due to the rapid spread of sites of reaction initiation because of the high thermal conductivity of the aluminum flakes.

Keywords: nanothermites; MICs; ball milling; mechanochemistry

Acknowledgments

The work was financially supported by the Basic Research Program of the Presidium of the Russian Academy of Sciences No. 35 "Scientific foundations for the creation of new functional materials" and projects of the Russian Foundation for Basic Research No. 16-03-00178a and No. 16-29-01030.

References

- Dreizin, E. 2009. Metal-based reactive nanomaterials. *Prog. Energ. Combust.* 35:141–167. doi: 10.1016/ j.pecs.2008.09.001.
- Dolgoborodov, A. Yu. 2015. Mechanically activated oxidizer-fuel energetic composites. *Combust. Explo. Shock Waves* 51(1):86–99.
- Dolgoborodov, A. Yu., A. N. Streletskiy, M. N. Makhov, V. A. Teselkin, Sh. L. Guseinov, P. A. Storozhenko, and V. E. Fortov. 2012. Promising energetic materials composed of nanosilicon and solid oxidizers. *Russ. J. Phys. Chem. B* 6(4):523–530.
- 4. Sundaram, D. S., P. Puri, and V. Yang. 2016. A general theory of ignition and combustion of nano- and micronsized aluminum particles. *Combust. Flame* 169:94–109. doi: 10.1016/j.combustflame.2016.04.005.
- Baijot, V., D. Mehdi, C. Rossi, and A. Esteve. 2017. A multi-phase micro-kinetic model for simulating aluminum based thermite reactions. *Combust. Flame* 180:10– 19. doi: 10.1016/j.combustflame.2017.02.031.
- Zakiyyan, N., A. Wang, R. Thiruvengadathan, C. Staley, J. Mathai, K. Gangopadhyay, M. R. Maschmann, and S. Gangopadhyay. 2018. Combustion of aluminum

nanoparticles and exfoliated 2D molybdenum trioxide composites. *Combust. Flame* 187:1–10. doi: 10.1016/j.combustflame.2017.08.027.

- Streletskii, A. N., M. V. Sivak, and A. Ju. Dolgoborodov. 2017. Nature of high reactivity of metal/solid oxidizer nanocomposites prepared by mechanoactivation: A review. J. Mater. Sci. 52:11810–11825. doi: 10.1007/s10853-017-1277-1.
- Sivak, M. V., A. N. Streletskiy, I. V. Kolbanev, A. V. Leonov, and E. N. Degtyarev. 2016. Thermal relaxation of defects in nanosized mechanically activated MoO₃. *Colloid J.* 78(5):674–684.
- Trunov, M. A., M. Schoenitz, X. Zhu, and Ed. Dreizin. 2005. Effect of polymorphic phase transformations in Al₂O₃ film on oxidation kinetics of aluminum powders. *Combust. Flame* 140:310–318. doi: 10.1016/ j.combustflame.2004.10.010.
- 10. Cabrera, N., and N. Mott. 1949. Theory of the oxidation of metals. *Rep. Prog. Phys.* 12:163.
- Grigor'ev, I. S., and E. Z. Meylikhov. 1991. *Fizicheskie velichiny: Spravochnik* [Physical quantities: Handbook]. Moscow: Energoatomizdat. 339 p.
- 12. Volkov, A.V., N.L. Kazanskiy, O.Yu. Moiseev, and S. D. Poletaev. 2015. Thermal oxidative degradation of

GORENIE I VZRYV (MOSKVA) - COMBUSTION AND EXPLOSION 2018 volume 11 number 1

molybdenum films under laser ablation. Tech. Phys. Russ. J. Appl. Phys. 60(2):265-269. doi: 10.1134/

S1063784215020255.

Received January 12, 2018

Contributors

Sivak Mikhail V. (b. 1985) — research scientist, N. N. Semenov Institute of Chemical Physics, Russian Academy of Sciences, 4 Kosygin Str., Moscow 119991, Russian Federation; sivak.mihail@gmail.com

Streletskii Andrey N. (b. 1945) — Doctor of Science in chemistry, head of laboratory, N. N. Semenov Institute of Chemical Physics, Russian Academy of Sciences, 4 Kosygin Str., Moscow 119991, Russian Federation; acting professor, Moscow Institute of Physics and Technology (State University), 9 Institutskiy Per., Dolgoprudny, Moscow Region 141701, Russian Federation; str@center.chph.ras.ru

Shevchenko Arseniy A. (b. 1991) — research engineer, N. N. Semenov Institute of Chemical Physics, Russian Academy of Sciences, 4 Kosygin Str., Moscow 119991, Russian Federation; Ph.D. student, National Research Nuclear University MEPhI (Moscow Engineering Physics Institute), 31 Kashirskoe Sh., Moscow 115409, Russian Federation; arsshevchenko@inbox.ru

Kolbanev Igor' V. (b. 1937) — senior research scientist, N. N. Semenov Institute of Chemical Physics, Russian Academy of Sciences, 4 Kosygin Str., Moscow 119991, Russian Federation; str@center.chph.ras.ru